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Use the chemical formula to find the number of atoms in one molecule and multiply this number by Avogadro's number, the

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number of particles in one mole. atom $6.02 \cdot 10^{23}$ O $2 \cdot 6.02 \cdot 10^{23}$ ion Na^+ $6.02 \cdot 10^{23}$ formula unit NaCl $6.02 \cdot 10^{23}$ $6.02 \cdot 10^{23}$ representative particles of a substance molecule formula unit atom

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10 Chemical Quantities91 SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER (pages 287–296) This section defines the mole and explains how the mole is used to measure matter. It

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