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2009, Prentice- Hall, Inc. Rate Laws • A rate law shows the relationship between the reaction rate and the concentrations of reactants. • The exponents tell the order of the reaction with respect to each reactant. • Since the rate law is Rate = k [NH₄⁺] [NO₂⁻] the reaction is. First-order in [NH₄⁺] and. First-order in [NO₂⁻].

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