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pairs of elements, underline the one that
would replace the other element in a
compound. a. calcium, tin e. iron, copper
b. bromine, fluorine f. iodine, chlorine c.
aluminum, potassium g. silver, lead d.
zinc, sodium 2.

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many. The most common use of this tool
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T102 Chemistry: Matter and Change

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Teaching Transparency Worksheets Chemistry Answers Chapter 19

TEACHING TRANSPARENCY WORKSHEET
Ionic Bonds Use with Chapter 7, Section 7.2

1. How many valence electrons does a neutral magnesium (Mg) atom have?
2. What is the charge on a magnesium ion? What does magnesium have to do to form such an ion, and why does it tend to do so?
3. How many valence electrons does a single neutral chlorine (Cl) atom have?
- 4.

Snow Elementary School - Dearborn Public Schools

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Chapter 3 Section 31 Worksheets - Leary Kids

Use with Chapter 7, Section 7.1 2. Are the elements shown on the left sides of the two equations neutral? How can you tell? 3. What is the name for the energy needed to remove electrons from an atom, such as the ... Teaching Transparency Worksheets Chemistry: ...

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Use the table of electronegativities below to answer the following questions. . Teaching Transparency Worksheets Chemistry: Matter and Change 0 Chapter 8 . 1.7, the CF bond is polar covalent.

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Use with Chapter 6, Section 6.2

MATHSKILLSTRANSPARENCY

WORKSHEET Using the Periodic Table 1.

Identify the number of valence electrons in each of the following elements. a. Ne d. Mg f. Cl h. Si 2. Identify the energy level of the valence electrons in each of the following elements. a. c. d. e. H Ar 1 3.

3.

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